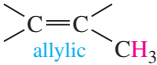
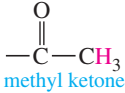
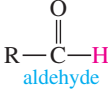
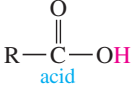
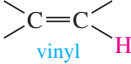


**TABLE 12-2** Summary of IR Stretching Frequencies

Frequency (cm <sup>-1</sup> )	Functional Group	Comments
3300	alcohol O—H amine, amide N—H alkyne $\equiv\text{C—H}$	always broad may be broad, sharp, or broad with spikes always sharp, usually strong
3000	alkane $\text{—}\overset{\text{H}}{\underset{\text{H}}{\text{C}}}\text{—H}$ alkene $\text{=}\overset{\text{H}}{\text{C}}\text{—}$ acid O—H	just below 3000 cm <sup>-1</sup> just above 3000 cm <sup>-1</sup> very broad
2200	alkyne $\text{—C}\equiv\text{C—}$ nitrile $\text{—C}\equiv\text{N}$	just below 2200 cm <sup>-1</sup> just above 2200 cm <sup>-1</sup>
1710 (very strong)	carbonyl $\text{>C=O}$	ketones, acids about 1710 cm <sup>-1</sup> aldehydes about 1725 cm <sup>-1</sup> esters higher, about 1735 cm <sup>-1</sup> conjugation lowers frequency amides lower, about 1650 cm <sup>-1</sup>
1660	alkene $\text{>C=C<}$ imine $\text{>C=N<}$ amide $\text{>C=O}$	conjugation lowers frequency aromatic C=C about 1600 cm <sup>-1</sup> stronger than C=C stronger than C=C (see above)

Ethers, esters, and alcohols also show C—O stretching between 1000 and 1200 cm<sup>-1</sup>.

**TABLE 13-3** Typical Values of Chemical Shifts

Type of Proton	Approximate $\delta$	Type of Proton	Approximate $\delta$
alkane ( $-\text{CH}_3$ ) methyl	0.9		1.7
alkane ( $-\text{CH}_2-$ ) methylene	1.3	Ph- <b>H</b> aromatic	7.2
alkane ( $-\text{CH}-$ ) methine	1.4	Ph- <b>CH</b> <sub>3</sub> benzylic	2.3
	2.1		9-10
$-\text{C}\equiv\text{C}-\text{H}$ acetylenic	2.5		10-12
R- $\text{CH}_2$ -X (X = halogen, O)	3-4	R-O <b>H</b> alcohol	variable, about 2-5
	5-6	Ar-O <b>H</b> phenol	variable, about 4-7
		R-N <b>H</b> <sub>2</sub> amine	variable, about 1.5-4

*Note:* These values are approximate, as all chemical shifts are affected by neighboring substituents. The numbers given here assume that alkyl groups are the only other substituents present. A more complete table of chemical shifts appears in Appendix 1.